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Structures and mechanical properties of Ti-Nb-Cr-V-Ni-Al refractory high entropy alloys

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Keywords: Refractory high entropy alloys Alloy design Microstructure Sigma phase Mechanical properties	Four refractory high entropy alloys with different chemical compositions, which can be calculated as $Ti_{(50.1.5625x)}Nb_{(30-0.9375x)}Cr_{10}V_{10}Ni_{1.5x}Al_x$ (x = 0, 5, 7, 10), were prepared by arc melting to determine the effect of Ni and Al on the phase composition, structure and mechanical properties. Each alloy was studied in both the as-cast and annealed at 1000 °C for 24 h conditions; compression tests at room temperature or at 800 °C was used to examine mechanical behavior of the alloys and the effect of deformation on microstructure. The replacement of Ti and Nb with Al and Ni resulted in the formation of the Ti, Ni-rich σ -phase and Ti ₂ Ni phases in the bcc matrix in contrast to the ThermoCalc predictions which suggests the formation of the bcc solid solution, B2, and Laves phases. The Ti, Ni-rich σ -phase can't also be expected from the analysis of the corresponding binary and ternary phase diagrams. Annealing did not result in noticeable changes in the microstructure. The formation of the σ and Ti ₂ Ni phases resulted in a considerable increase in strength (from 745 to 1600 MPa) and decrease in ductility (from a thickness reduction >50% to fracture in the elastic region) at room temperature. An increase in the

1. Introduction

The so-called refractory high entropy alloys (RHEAs) were introduced recently as a new class of metallic alloys potentially suitable for high-temperature applications [1,2]. Widely used Ni-based superalloys cannot satisfy future needs for high-temperature materials, for example, in the aerospace sector, due to the limited incipient melting temperature [3]. RHEAs and related refractory complex concentrated alloys (RCCAs) are multi-component (generally the number of components is \geq 5) close to equiatomic alloys composed of elements with high melting temperatures like Cr, Hf, Mo, Nb, Ta, Ti, V, W and Zr; the presence of non-refractory elements like Al, Co, Ni, and Si is also possible [2,4]. RHEAs and RCCAs with impressive strength at temperatures up to 1600°C and promising oxidation resistance have already been reported [2,5–18]. However, further optimization of composition, structure, and properties of such alloys is still required to develop materials attractive for practical applications.

Mechanical properties of the metallic alloys are primarily dependent

on their structure and phase composition. RHEAs mainly consist of a random base centered cubic (bcc) solid solution and a B2 ordered phase, also based on the bcc lattice [2]. The B2 phase is often observed in alloys with Al [19–23]. Some alloys were revealed to have a single B2 phase structure [20,23,24], sometimes containing a small amount of non-coherent secondary phases like a Laves phase, σ -phase or Zr aluminides [25,26].

testing temperature to 800 °C resulted in softening to and a substantial increase in ductility of all the alloys. Complex relationships between the fraction of the second phases and mechanical properties were discussed.

However the most attractive option is probably associated with the production of a mixture of the bcc + B2 coherent phases. In some alloys cuboidal or plate-like bcc precipitates were found in a continuous B2 matrix; this type of microstructure was sometimes referred to as a "superalloy"-like structure [21,22] and these alloys can be termed as "refractory high entropy superalloys" (RHESAs) [22]. Some of the RHESAs has already demonstrated high specific strength at $T \leq 1200$ °C, significantly outperforming properties of conventional Ni-based superalloys [1,22]. However, RHESAs generally possess limited ductility at low temperatures, mostly due to the ordered nature of the matrix phase. Some efforts to "invert" the structure of the available RHESAs into the

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bcc matrix with B2 precipitates by proper alloying and/or heat treatment [27], did not clarify thoroughly the accessibility of this goal.

Nevertheless, the idea to obtain a better combination of mechanical properties through the formation of coherent precipitates in the bcc solid solution remains very promising. In high entropy alloys based on transition metals (TMHEAs) the formation of coherent particles in a stable solid solution matrix can be attained through the addition of the proper alloying elements [28–38].

To extend this approach to RHEAs, a $Ti_{50}Nb_{30}Cr_{10}V_{10}$ alloy with a single bcc solid solution phase structure developed earlier [39,40] was used as the reference point. The $Ti_{50}Nb_{30}Cr_{10}V_{10}$ alloy was then doped with Ni and Al (the contents of Ti and Nb were proportionally decreased). The following reasons were taken in consideration: (i) binary NiAl has the B2 structure presumably coherent with the bcc matrix; (ii) the enthalpy of the intermetallic phase formation in the Ni-Al pair is much more negative in comparison with any other possible pair ((Ti, Nb, Cr, V)-Al) [41]; i.e. the formation of NiAl is most preferable among all other Al-containing phases. A preliminary assessment of the phase composition of proposed alloys by the CALPHAD method (see Fig. 1) suggested also the formation in some of them the B2 particles in the bcc matrix. Different amounts of Ni and Al will presumably produce different fractions of second phase particles in the bcc matrix.

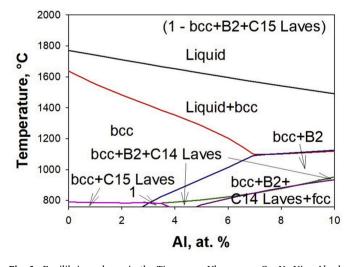
To determine the effect of Ni and Al experimentally, a set of alloys with different chemical compositions (which can be calculated as $Ti_{(50-1.5625x)}Nb_{(30-0.9375x)}Cr_{10}V_{10}Ni_{1.5x}Al_x$ (x = 0, 5, 7, 10)) was prepared by arc melting. Structure and compression mechanical properties of the alloys at room and elevated (800 °C) temperatures were investigated.

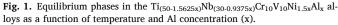
2. Materials and methods

The $Ti_{50}Nb_{30}Cr_{10}V_{10}$ (referred hereafter to as x = 0), $Ti_{42.2}Nb_{25.3}Cr_{10}V_{10}Ni_{7.5}Al_5$ (x = 5), $Ti_{39.1}Nb_{23.4}Cr_{10}V_{10}Ni_{10.5}Al_7$ (x = 7) and $Ti_{34.4}Nb_{20.6}Cr_{10}V_{10}Ni_{15}Al_{10}$ (x = 10) alloys were produced by arc melting of the elements in a low-pressure, high-purity argon atmosphere inside a water-cooled copper cavity. The ingots were remelted at least 5 times to ensure chemical homogeneity; the alloying elements purity was no less than 99.9 wt%.

The produced ingots of the alloys had dimensions of about $6 \times 12 \times 40 \text{ mm}^3$. Specimens measured $4 \times 4 \times 6 \text{ mm}^3$ were cut from the ingots using an electric discharge machine. Some of the specimens were annealed at 1000 °C for 24 h. Prior to annealing, the specimens were sealed in vacuumed (10^{-2} Torr) quartz tubes filled with titanium chips to prevent oxidation.

The microstructure of the alloys in the as-cast and annealed at 1000





°C conditions were studied using X-ray diffraction (XRD), scanning electron microscopy (SEM) and electron backscatter diffraction (EBSD) analysis. Specimens for XRD, SEM, and EBSD were produced by careful mechanical polishing. XRD analysis was performed using a RIGAKU diffractometer with CuKa radiation. SEM investigations were performed on an FEI Quanta 600 field emission gun (FEG) microscope equipped with an energy-dispersive (EDS) detector. EBSD was conducted using an FEI Nova NanoSEM 450 FEG SEM equipped with a Hikari EBSD detector and a TSL OIMTM system version 6.0. This software was used to generate inverse pole figures (IPF) and phase maps. On the presented IPF maps the high angle boundaries (HABs) are indicated with black lines while white lines denote the low angle boundaries (LABs). The bcc, σ , and Ti₂Ni phases are shown in the presented phase maps with red, green, and blue colors, respectively. All the EBSD maps are shown in the asobtained conditions without any additional cleanup procedures. The volume fractions of different phases were measured by the areal analysis of SEM-BSE images using a Digimizer Image Analysis Software and EBSD phase maps. The difference between the fractions of the phases measured from SEM-BSE and EBSD data for the same alloy and condition was <2%; the average value is given.

Nanohardness was measured using a Shimadzu DUH-211s Dynamic Ultra Micro Hardness Tester with a Berkovich tip. At least twenty indents were performed on each phase avoiding any influence from other phases. The maximum load of 50 mN was applied for 5 s; a loading speed was 6.662 mN/s. The hardness was calculated using the Oliver and Pharr method [42].

Isothermal compression of the as-cast and annealed at 1000 °C rectangular specimens measured 4 × 4 × 6 mm³ was carried out at 22 or 800 °C using an Instron 300LX testing machine equipped with a radial furnace. The specimens were placed into the preheated to the testing temperature furnace and held for ~10 min to equilibrate the temperature prior to testing. The temperature was controlled by a thermocouple attached to a side surface of the specimen. The initial strain rate was 10⁻⁴ s⁻¹.

The equilibrium phase diagram was constructed using a Thermo-Calc (version 2019b) software and a TCHEA3 (high-entropy alloys) database.

3. Results

3.1. Phase diagram

Fig. 1 shows equilibrium phases in the $Ti_{(50\cdot1.5625x)}Nb_{(30\cdot0.9375x)}Cr_{10}V_{10}Ni_{1.5x}Al_x$ alloys as a function of temperature and Al concentration (x). Phases stable at 1000 °C, their chemical composition and fractions in the alloys with $x=0,\,5,\,7$ or 10 are given in Table 1. One can clearly see that the phase composition of the alloys strongly depended on the Al concentration. The "base" $Ti_{50}Nb_{30}Cr_{10}V_{10}$ alloy was solidified in a temperature interval of 1637–1770°C with the formation of a bcc solid solution upon solidification. Only the bcc phase remained stable during cooling till 790 °C. At 790 °C, a Cr, Nb-rich C15 (fcc) Laves phase precipitated.

The addition of Al (accompanied also by an increase in the Ni content at the expense of Ti and Nb) had the following effects:

Table 1

Stable phases and their chemical compositions at 1000 $^\circ C$ in the $Ti_{(50-1.5625x)}Nb_{(30-0.9375x)}Cr_{10}V_{10}Ni_{1.5x}Al_x$ (x = 0, 5, 7, 10) alloy according to the Thermo-Calc software.

Alloy	Stable phases	Ti	Nb	Cr	v	Ni	Al	$V_{\rm f}(\%)$
$\mathbf{x} = 0$	BCC	48.8	31.2	10.0	10.0	-	-	100.0
x = 5	BCC	42.5	25.0	10.0	10.0	7.5	5.0	100.0
$\mathbf{x} = 7$	BCC	40.0	23.0	10.2	10.2	9.0	7.6	96
	B2	40.2	10.2	4.5	6.8	32.5	5.8	4
x = 10	BCC	34.3	23.8	12.0	11.3	8.3	10.3	72
	B2	36.7	10.8	5.2	6.8	31.1	9.4	28

- (i) A pronounced decrease in both the liquidus and solidus temperatures and some broadening of the solidification range. For example, the liquidus temperature monotonically decreased from 1770 °C to 1490 °C with an increase in x from 0 to 10. The solidus temperature, meanwhile, first decreased from 1637°C to 1098°C with an increase in x from 0 to 7, and then increased slightly to 1119°C at x = 10.
- (ii) A Ni-rich B2 phase started to appear at x = 2.7. Its solvus temperature increased quickly with an increase in the Al concentration. The B2 solvus temperature approached 1099°C and became slightly higher than the solidus temperature of the alloy.
- (iii) The C15 (fcc) Laves phase transformed into a C14 (hexagonal) one at x = 3.5–3.8. Note that a similar C15 to C14 transformation was already demonstrated experimentally in some Al-Cr-Nb-Ti-V-Zr RHEAs with an increase in the content of Al [15]. The C14 Laves phase was enriched with Cr and Nb, the solvus temperature of this phase increased from 790°C to 953°C with an increase in x from 4 to 10.
- (iv) Finally, a Ni, Ti-rich fcc phase appeared at x = 4.8; the solvus temperature of this phase increased quickly approaching 953 $^\circ C$ at x = 10.

3.2. Structure of the as-cast alloy

The effect of Al and Ni content on the structure of the program Ti-Nb-Cr-V-Ni-Al alloys in the as-cast condition was estimated by XRD (Fig. 2). The "starting" $Ti_{50}Nb_{30}Cr_{10}V_{10}$ alloy had the single bcc phase structure with the lattice parameter of 0.3222 nm (lattice parameters of constitutive phases are given in Supplementary information, Table S1). The effect of Ni and Al was associated with the appearance of additional phases, namely a cubic Ti_2Ni phase (the space group 227) and a tetragonal σ -phase (the space group 136). The lattice constant of the bcc phase reduced slightly to 0.3175–0.3179 nm when the content of Ni and Al increased. Some increase in the intensity of the σ -phase peaks and a decrease in the intensity of peaks from the bcc phase at high Al/Ni concentrations should be noted.

The addition of Al (and Ni) had a significant effect on the as-cast microstructure of the Ti-Nb-Cr-V-Ni-Al alloys (Fig. 3). The $Ti_{50}Nb_{30}Cr_{10}V_{10}$ alloy had a single bcc phase structure, as confirmed by EBSD results (Fig. 3b). The average bcc grain size was $140 \pm 100 \mu$ m; however both quite fine (~50 µm) and very coarse (~500 µm) grains were found. Dendritic segregations were observed inside the bcc grains (Fig. 3a). Lighter dendritic areas (#1 in Fig. 3a) were enriched with Nb,

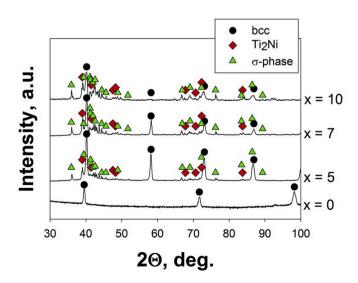


Fig. 2. XRD pattern of the Ti-Nb-Cr-V-Ni-Al alloys with different Al contents (x) in the as-cast state.

darker interdendritic areas (#2) were found to be enriched with Ti, Cr, and V (Table 2).

Relatively small amounts of Al and Ni in the x = 5 alloy resulted in a more complex microstructure of the alloy (Fig. 3c, d). An EBSD phase map (Fig. 3d) showed the formation of a substantial amount (13%) of the σ -phase (#3 in Fig. 3c) in the bcc phase matrix (#1 – dendrites and #2 - interdendrites). The average bcc grain size was 20 ± 9 µm; the elongated σ particles had an irregular polygonal shape with the average width of $3\pm1~\mu m$. A small amount of the Ti₂Ni phase (<1%) was also detected (#4); the Ti₂Ni particles are visible as black spots in the SEM-BSE images. Both the σ and Ti₂Ni phases were predominantly located in the interdendritic areas (Fig. 3c), which were enriched with Ti and Cr (Table 2). The chemical composition of the Ti₂Ni particles was rather close to the nominal stoichiometry (including, however, \sim 3–6 at.% of the rest of the constitutive elements of the alloy). In turn, the σ -phase had a complex chemical composition which can be described as (Ti, Cr, Nb, V)_{0.7}(Ni, Al)_{0.3}, with the dominant presence of Ti (44.5 at.%) and Ni (22.5 at.%).

An increase in the percentage of Al in the x = 7 alloy did not result in drastic changes in the microstructure (Fig. 3e, f). However, the fraction of both the σ (#2 in Fig. 3e) and Ti₂Ni (#3) phase particles increased substantially to 25% and 3%, respectively. The average size of the σ and Ti₂Ni particles was 4±2 µm and 1.0 ± 0.7 µm, respectively. Note that the interdendritic area, which formed in the x = 7 alloy a continuous network, seemed to be fully consumed by the second phases, and only the dendritic bcc areas remained (#1). The chemical compositions of the σ and Ti₂Ni phases were close to those in the previous alloy while the bcc phase became depleted of Ni and enriched in Nb (Table 2).

Even more complex microstructure was found in the alloy with the maximum amount of Al (x = 10) (Fig. 3g, h). Microstructure has a typical dendritic morphology, however only the dendritic areas were occupied by the bcc phase with the volume fraction of 31% (#1 in Fig. 3g), while the interdendritic areas consisted of both the σ -phase (60%, #2) and Ti₂Ni (9%, #3) phase. The σ phase particles formed a continuous network with discrete areas of the bcc and Ti₂Ni phases. The chemical composition of the phases was similar to those in the previous alloys (Table 2).

3.3. Structure of the alloys after annealing (1000 °C, 24 h)

Fig. 4 shows XRD patterns of the Ti-Nb-Cr-V-Ni-Al alloys with different Al content (x) after annealing at 1000°C for 24 h. One can clearly see that the annealing treatment did not change the phase composition of the alloys significantly: the $Ti_{50}Nb_{30}Cr_{10}V_{10}$ alloy had the single bcc phase structure, while the addition of Al and Ni resulted in the appearance of the σ and Ti_2Ni phases. The intensity of the bcc peaks decreased with an increase in the Al/Ni content. The lattice parameters of the phases were slightly affected by annealing (Supplementary information, Table S2). For example, the lattice constant of the bcc phase changed from 0.3225 nm in the x = 0 alloy to 0.3181–0.3185 nm in the x = 5–10 alloys (Table 4). The lattice parameters of the σ and Ti_2Ni phases also varied in a very limited interval (see Tables S1 and S2).

Both SEM-BSE images and EBSD IPF/phase maps of the alloy structure after annealing treatment are shown in Fig. 5. The $Ti_{50}Nb_{30}Cr_{10}V_{10}$ alloy had a single-phase granular microstructure with a polygonal shape of the grains (Fig. 5a, b). The dendritic segregations were almost not observed in the microstructure (Fig. 5a). The average bcc grain size was $170\pm80~\mu\text{m}$; the grain size became coarser while the scatter became lower in comparison with the as-cast state (Fig. 3a, b). The chemical composition of the bcc grains was close to the nominal composition of the alloys (Table 3).

Fig. 5c, d shows the microstructure of the x = 5 alloy. The dendritic segregations in the bcc matrix disappeared after annealing (Fig. 5c). Inside the bcc matrix (#1 in Fig. 5c), coarse, irregularly shaped particles of the σ -phase were found (Fig. 5d). These particles with the size of 4 \pm 1 μ m are dark-grey in the SEM-BSE image (#2); the fraction of the

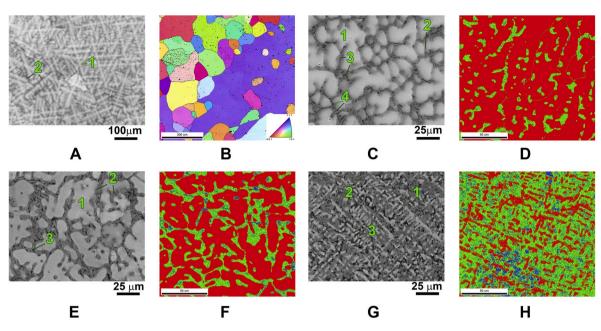


Fig. 3. Microstructure of the Ti-Nb-Cr-V-Ni-Al alloys with different Al contents in the as-cast state: a, b - x = 0; c, d - x = 5; e, f - x = 7; g, h - x = 10; a, c, e, g - SEM-BSE images; b - IPF map; d, f, h - phase maps (red color – bcc; green – σ ; blue – fcc (Ti₂Ni) phases). The chemical compositions of the constitutive phases are shown in Table 2. (For interpretation of the references to color in this figure legend, the reader is referred to the Web version of this article.)

Table 2

Chemical compositions of the structural constituents of the program Ti-Nb-Cr-V-Ni-Al alloys with different Al content (x) in the as-cast state. The corresponding phases are shown in Fig. 3.

Elemer	nts, at.%	Ti	Nb	Cr	V	Ni	Al	Vf (%)	
Constit	ruents	$\mathbf{x} = 0$							
N [⊙]	Designation								
1	Bcc dendrite	46.6	37.2	7.1	9.1	-		-	100
2	Bcc interdendrite	55.0	18.5	14.9	11.6	-		-	
Alloy c	composition	50.0	28.9	10.5	10.6	-		-	-
Nomin	al composition	50	30	10	10	-		-	-
		x = 5							
1	Bcc dendrite	41.5	29.2	9.6	11.0	3.4	5.3	86	
2	Bcc interndendrite	44.3	22.7	11.6	10.9	4.9	5.6		
3	σ-phase	44.5	10.8	9.0	5.9	22.5	7.3	13	
4	Ti2Ni	52.4	6.6	4.8	3.4	26.4	6.4	<1	
Alloy c	composition	43.3	22.8	10.5	9.8	7.9	5.7	-	
Nomin	al composition	42.2	25.3	10.0	10.0	7.5	5	-	
		$\mathbf{x} = 7$							
1	Bcc phase	40.4	23.0	11.3	12.0	5.2	8.1	72	
2	σ-phase	39.4	11.9	10.3	6.9	21.9	9.6	25	
3	Ti2Ni	52.1	7.3	4.9	3.8	25.6	6.3	3	
Alloy c	composition	42.1	20.1	9.9	9.7	10.4	7.8	-	
Nomin	al composition	39.1	23.4	10.0	10.0	10.5	7.0	-	
		x = 10							
1	Bcc phase	32.3	26.0	11.7	13.7	5.8	10.5	31	
2	σ-phase	31.7	16.1	10.8	8.7	20.4	12.3	60	
3	Ti2Ni	43.3	10.3	6.7	5.9	23.2	10.6	9	
Alloy c	composition	34.8	18.7	9.9	9.8	15.6	11.2	-	
Nomin	al composition	34.4	20.6	10.0	10.0	15.0	10.0	_	

 σ -phase was 11%. Fine black particles (#3) can be found inside the σ particles or at the σ/bcc interfaces (Fig. 5c); these black particles can be identified as the Ti₂Ni phase based on the result of EBSD analysis (Fig. 5d). The fraction and average size of the Ti₂Ni particles was 2% and 2.0 ± 0.7 µm, respectively. The chemical compositions of the constitutive phases (Table 3) were quite close to those of the as-cast alloy (Table 2).

An increase in the Al content in the x = 7 alloy did result in an noticeable increase in the fraction of the second phases (Fig. 5e, f): the percentages of the σ and Ti₂Ni phases were 29% and 3%, respectively. The σ phase particles (#2 in Fig. 5e) usually formed characteristic cellular conglomerates, however individual particles can also be found.

The Ti₂Ni phase particles (#3) were located either inside the σ particles or at the σ/bcc interfaces. The transversal sizes of the σ particles and bcc grains were 5 \pm 1 μm and 2.0 \pm 0.6 μm , respectively. The chemical compositions of the phases are given in Table 3.

A continuous network of the σ phase (#2 in Fig. 5h) with a fraction of 65% was found in the alloy with 10 at.% of Al (Fig. 5g, h). The bcc particles (#1) with a fraction of 23% and the average size of 8 ± 4 µm were found inside the σ matrix. In addition, relatively coarse (in comparison with the alloys with a lower content of Al) Ti₂Ni particles (#3) with the size and fraction of 5 ± 3 µm and 12%, respectively, were observed. The chemical compositions of the phases were similar to those in the alloys with a lower content of Al and were the same as in the as-

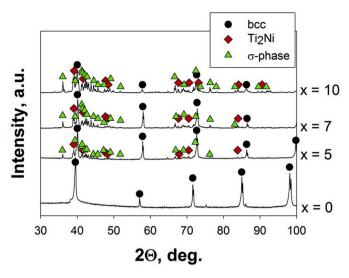


Fig. 4. XRD pattern of the Ti-Nb-Cr-V-Ni-Al alloys with different Al content (x) after annealing at 1000 $^\circ C$ for 24 h.

cast alloy with 10 at.% of Al (Tables 2 and 3).

3.4. Mechanical properties

Mechanical properties of the Ti-Nb-Cr-V-Ni-Al alloys in the as-cast and annealed conditions were evaluated during compression at room temperature or at 800 °C. In both conditions the alloys demonstrated very similar mechanical behavior, thus only data for the annealed condition is presented here. The stress-strain curves obtained at room temperature and 800°C are shown respectively in Fig. 6a and b, while the obtained characteristics (yield strength (σ_{YS}), peak strength (σ_p), and compressive strain (ϵ)) are summarized in Table 4. The data for the as-cast condition can be found in Supplementary information (Fig. S1 and Table S3).

The annealed $Ti_{50}Nb_{30}Cr_{10}V_{10}$ alloy had relatively low yield strength at room temperature (755 MPa) however due to good strain hardening capacity the alloy did not fracture even after reaching 50% of height reduction (Fig. 6a). The addition of 5 at.% of Al resulted in a slight increase in the yield strength to 775 MPa, while the peak strength was found to be 1240 MPa. Meanwhile ductility of the alloy decreased to 13.8% in this alloy. A considerable improvement in strength was obtained in the alloy with 7 at.% of Al. In this case, the yield strength of 1600 MPa was attained together with reasonable ductility of 8.9%. Note that the apparent work hardening rate (i.e. the slope of the plastic deformation domain of the curve) all of the x = 0-7 alloys was almost identical. Meanwhile further increase in the Al concentration resulted in a complete loss in ductility; the x = 10 alloy has fractured in the elastic region at 1830 MPa.

An increase in the testing temperature to 800 °C resulted in considerable softening and a substantial improvement in ductility of the studied Ti-Nb-Cr-V-Ni-Al alloys (Fig. 6b). For example, the yield strength was 45, 85, 110, or 285 MPa for the alloys with x = 0, 5, 7 or 10, respectively. All the alloys were deformed in compression to 50% of height reduction without fracture. However, shapes of the stress-strain curves differed from each other considerably depending on the Al concentration. The alloys with x = 0-7 demonstrated after yielding an extended stage nearly steady-state flow stage. Note that the x = 7 alloy showed some signs of hardening at reduction >10%. The x = 10 alloy softened continuously during the plastic flow after yielding.

To get an additional insight into the mechanical behavior of the program alloys, nanohardness of the constitutive phases was measured in the annealed x = 10 alloy (Table 5). The bcc phase was found to be the softest one with the hardness of 530 HV, while the Ti₂Ni phase was significantly harder (935 HV). However, the highest hardness demonstrated the σ -phase (1085 HV).

Table 4

Compressive mechanical characteristics (σ_{YS} - yield strength, σ_{YS} - peak strength, ϵ – compressive strain) of the Ti-Nb-Cr-V-Ni-Al alloys with different Al content (x) after annealing at 1000 °C for 24 h obtained at RT or 800 °C.

Alloy	σ _{YS} , MPa	σ _p , MPa	ε, %
RT			
$\mathbf{x} = 0$	755	_	\geq 50
x = 5	775	1135	13.8
$\mathbf{x} = 7$	1270	1685	8.9
x = 10	_	1395	0
800 °C			
$\mathbf{x} = 0$	45	60	\geq 50
x = 5	85	105	\geq 50
$\mathbf{x} = 7$	110	200	\geq 50
x = 10	285	395	\geq 50

Table 3

Chemical compositions of the structural constituents of the program Ti-Nb-Cr-V-Ni-Al alloys with the different Al content (x) after annealing at 1000 °C for 24 h. The corresponding phases are shown in Fig. 5.

Elements, at.%)	Ti		Nb		Cr		V	Ni	Al	Vf (EBSD) %
Constituents		$\mathbf{x} = 0$									
N [○] D€	esignation										
Bcc phase	0	49.6	29.8		10.3		10.3		_		
Alloy composit	tion	50.0	28.9		10.5		10.6		_		_
Constituents		$\mathbf{x} = 5$									
Nº	Designation										
1	Bcc phase	44.1		24.3		10.1		10.2	5.5	5.8	87
2	σ-phase	37.0		14.1		14.1		8.2	20.4	6.2	11
3	Ti2Ni	59.6		3.2		3.4		0.7	29.8	3.3	2
Alloy composition		43.3		22.8		10.5		9.8	7.9	5.7	_
		$\mathbf{x} = 7$									
1	Bcc phase	42.9		23.2		9.3		10.3	5.5	7.7	68
2	σ-phase	35.2		14.4		12.8		8.0	20.6	9.0	29
3	Ti2Ni	57.8		4.0		3.6		1.1	28.8	4.7	3
Alloy composit	tion	42.1		20.1		9.9		9.7	10.4	7.8	_
		x = 10									
1	Bcc phase	38.9		29.4		8.3		12.8	4.8	5.8	23
2	σ-phase	33.7		16.4		12.3		10.0	21.1	6.5	65
3	Ti2Ni	59.3		3.5		2.7		0.9	30.2	3.4	12
Alloy composit	tion	34.8		18.7		9.9		9.8	15.6	11.2	_

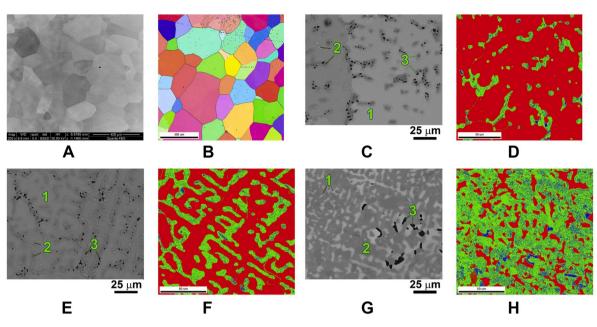


Fig. 5. Microstructure of the Ti-Nb-Cr-V-Ni-Al alloys with different Al content (x) after annealing at 1000 °C for 24 h: a, b - x = 0; c, d - x = 5; e, f - x = 7; g, h - x = 10; a, c, e, g – SEM-BSE images; b – IPF map (the color code is shown in Fig. 3b); d, f, h – phase maps (red color – bcc; green – σ ; blue – fcc (Ti₂Ni) phases). The chemical composition of the constitutive phases is shown in Table 3. (For interpretation of the references to color in this figure legend, the reader is referred to the Web version of this article.)

3.5. Structure after deformation at 800 °C

Structures of the x = 0 and x = 10 alloys after compression to 50% height reduction at 800 °C are shown in Fig. 7. XRD did not reveal any changes in the phase composition of the alloys after hot deformation (Fig. 7a): the x = 0 alloy was composed of the single bcc phase and x =10 alloy contained a mixture of the bcc, σ , and Ti₂Ni phases. The deformed microstructure of the x = 0 alloy consisted of characteristic pancake-shaped grains (Fig. 7b). Inside some grains, a well-developed substructure without any signs of dynamic recrystallization was observed. In the x = 10 alloy the bcc phase (#1 in Fig. 7c), most probably acting as a "soft" layer between hard particles of the σ -phase, accommodated a large amount of plastic strain. However, the morphology of the σ -phase particles (#2) also implies their participating in the plastic flow. The characteristic continuous network of the σ -phase in the initial (annealed at 1000°C (Fig. 5g and h)) condition broke during deformation so that individual σ particles aligned with the metal flow direction and separated by the bcc layers were found. The Ti₂Ni particles (#3) seemed to be not affected significantly by the deformation.

4. Discussion

4.1. Phase composition and microstructure of the Ti-Nb-Cr-V-Ni-Al alloys

The obtained results show a significant effect of the chemical composition (i.e. a decrease in the concentration of Ti and Nb and an increase in the content of Ni and Al in the $Ti_{(50-1.5625x)}Nb_{(30-0.9375x)}Cr_{10}V_{10}Ni_{1.5x}Al_x$ (x = 0, 5, 7, 10) alloys) on microstructure of the program alloys in the as-cast and annealed (1000 °C) conditions. An increase in x resulted in the formation of the Ti, Ni-rich σ , and Ti₂Ni phases rather than the B2 Ni-rich phase which was anticipated from the enthalpies of the intermetallic phases formation [41] and CALPHAD predictions (Fig. 1). The additional phases formation can be expected since Ti and Nb are known to be bcc phase stabilizers in RHEAs [2,43], while Al is a compound former, at least in certain combinations of elements [43–45]. However, no experimental information on the effect of Ni on the structure and properties of RHEAs was available before.

The equilibrium phase diagram, constructed for the Ti(50-1.5625x)Nb(30-0.9375x)Cr₁₀V₁₀Ni_{1.5x}Al_x with x = 0-10 (Fig. 1) suggested the bcc solid solution formation during solidification with possible precipitation of the fcc solid solution, Ni-rich B2 phase, and C14/C15 Laves phases during further cooling. The solvus temperatures of the fcc solid solution and the Laves phase were well below 1000 °C. Therefore, it is not surprising that these phases were not found in the as-cast state and after annealing at 1000 °C. The solvus temperature and the fraction of the B2 phase quickly increased with an increase in x, and the alloys with x > 7 were supposed to contain B2 phase in solid-state always. However, the B2 phase was also not detected even in the alloy with x = 10. Instead the σ -phase and a small portion of the Ti₂Ni phase appeared in the x = 5 alloy; their fraction increased with a further increase in x (Figs. 2-5). Thus, the ThermoCalc calculations correctly predicted a single bcc solid solution structure of the Ti₅₀Nb₃₀Cr₁₀V₁₀ alloy (it worth noting that no additional phases were found in the alloy after deformation at 800°C (Fig. 8a) which is also consistent with the phase diagram) and the formation of a second phase(s) with the addition of Al and Ni; however the exact type of these phases cannot be reliably determined by this method.

It is already well established that the accuracy of the CALPHAD predictions is critically dependent on the validity of the thermodynamic databases used [46,47]. A full thermodynamic description of each binary and ternary phase diagram belonging to the multicomponent system is required for the accurate calculation. For example, the senary Al-Cr-Nb-Ni-Ti-V system contains 15 binary and 20 ternary systems. The TCHEA3 database used in the current study contained a full description of 15 binaries for this system [48] which means that the fraction of the accessed binaries (FAB; [46]) was equal to 1. However, out of 20 ternaries, only 6 were fully accessed, and 11 were accessed fully or tentatively. This means that the fraction of the accessed ternaries (FAT; [46]) was 0.3–0.55. Given relatively low accuracy of the predicted phase diagram in the current case (Fig. 1), one can conclude that a high FAB is not enough because a higher FAT is required to get more accurate predictions, which is in agreement with some recent results [49].

Since it was not predicted by the Thermo-Calc software, the available information from binary and ternary phase diagrams [50,51] was also analyzed to get more clues on the σ -phase formation. Among the 15

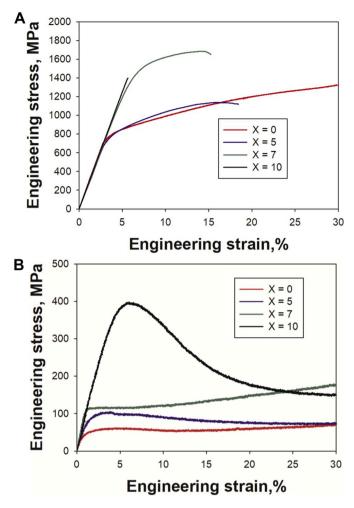


Fig. 6. Engineering stress-strain curves obtained during compression at RT (a) or at 800 $^{\circ}$ C (b) of the Ti-Nb-Cr-V-Ni-Al alloys with different Al content (x) after annealing at 1000 $^{\circ}$ C for 24 h.

Table 5

Nanohardness of the constitutive phases of the $Ti_{34,4}Nb_{20,6}Cr_{10}V_{10}Ni_{15}Al_{10}~(x=10)$ alloy after annealing at $1000^\circ C$ for 24 h.

Phase	Hanohardness, HV
bcc	530 ± 20
Ti ₂ Ni	935 ± 70
σ-phase	1085 ± 25

binaries, only 2 contained information about the σ-phase: Al-Nb and Ni-V. In the Al-Nb system the σ -phase had quite a narrow homogeneity region around 70% of Nb (AlNb₂), while in the Ni-V system the V concentration was in the range of 55–75% (NiV₂). In the ternary systems, 8 phase diagrams contained information on the σ -phase: Al-Cr-Nb, Al-Nb-Ni, Al-Nb-Ti, Al-Nb-V, Al-Ni-V, Cr-Ni-V, Nb-Ni-V and Ni-Ti-V. All 8 ternaries included either Al-Nb or Ni-V binary. The AlNb₂ σ -phase was able to dissolve up to ~30 at.% of Ti or ~20% of V, while the NiV₂ σ -phase generally dissolved only ~ 10 at.% of a third component. It also should be mentioned that the existence of the σ -phase in the Cr-Ni binary system was sometimes reported [52], yet these claims were criticized therefore the σ -phase is not believed to be an equilibrium phase in the Cr-Ni system [53]. Also note that the Ti₂Ni phase is present in the Ni-Ti binary and the Ni-Ti-Y (Y=Al, Cr, Nb, V) ternary phase diagrams. These phase diagrams suggest quite a narrow homogeneity range which agrees with the chemical composition close to the stoichiometric one found experimentally in the present Al-Cr-Nb-Ni-Ti-V allovs (Tables 2 and 3).

The binary and ternary phase diagrams suggested that the σ -phase in the Al-Cr-Nb-Ni-Ti-V alloys can be either of an AlNb₂ or NiV₂ type; the AlNb₂ type σ-phase has already been reported in some Al-Cr-Nb-Ti-V RHEAs [16.25]. However. in the examined $Ti_{(50\text{-}1.5625x)}Nb_{(30\text{-}0.9375x)}Cr_{10}V_{10}Ni_{1.5x}Al_x$ (x = 5, 7, 10) the $\sigma\text{-}phase$ was Ti, Ni-rich which is approximately corresponding to the (Ti, Cr, Nb, V)_{0.7}(Ni, Al)_{0.3} stoichiometry. This phase: (i) cannot be anticipated from the Ni-Ti binary and Ni-Ti-Y (Y=Al, Cr, Nb, V) ternaries; (ii) was never reported in the literature. The formation of the Ti, Ni-based σ -phase in the program alloys can be attributed to both the intricacies of chemical composition and complex interaction of the constitutive elements that stabilize new phases, unexpected from the lower order phase diagrams. However, additional experiments are required to establish the stability of the observed Ti, Ni-rich σ -phase.

An increase in the Al (and Ni) percentage had resulted in a considerable increase in the fraction of the Ti, Ni-rich σ -phase. For instance, the amount of the σ particles was 11–13% and 60–65% in the x=5 and x=10 alloys, respectively (Tables 2 and 3). Furthermore, the sum of the atomic percentages of Al and Ni in the σ particles was almost constant (29.8-32.7 at.% in the as-cast condition and 26.6-29.6 at.% in the annealed condition (Tables 3 and 5)) irrespective of the composition of the alloys. In addition, the total concentration of Al and Ni in the bcc phase in the annealed condition was also almost constant (10.6–13.2 at. %). However, in the as-cast condition much more significant scatter was found (from 9.6 at.% in the x = 5 alloy to 16.3 at.% in the x = 10 alloy) most probably due to the non-equilibrium nature of the phases produced during solidification. It therefore seems that the x = 5 composition (i.e. 5 at.% of Al and 7.5 at.% of Ni) is very close to the border between the bcc and bcc+ σ phase fields and the alloys with a lower amount of Al and Ni should most probably have a single bcc phase structure.

Since the solubility of Al and Ni in both the bcc and σ phases is fixed

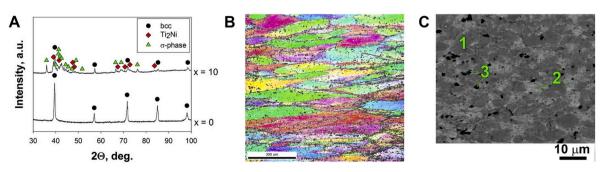


Fig. 7. Structure of the Ti-Nb-Cr-V-Ni-Al alloys with different Al content (x) after annealing at 1000 °C for 24 h and compression at 800 °C: (a) XRD patterns of the x = 0 and x = 10 alloys; (b) EBSD IPF map of the x = 0 alloy (the color code is shown in Fig. 3b); (c) SEM-BSE image of the x = 10 alloy (bcc phase is identified with #1, σ -phase – with #2, Ti₂Ni phase – with #3). The compression axis is vertical. (For interpretation of the references to color in this figure legend, the reader is referred to the Web version of this article.)

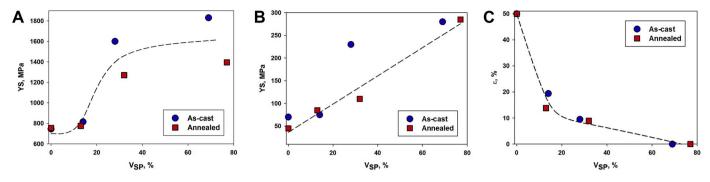


Fig. 8. Dependence of the yield strength (a, b) at RT (a) and 800° C (b) and ductility (c) of the program Ti-Nb-Cr-V-Ni-Al alloys on the volume fraction of the second phases. Dashed lines are only guides to eyes. In the case of x = 10 alloy the peak strength was used instead of the yield strength at RT.

(at least in a rather equilibrium annealed condition), the amount of the σ -phase in the alloy is ought to be controlled by the total amount of Al and Ni in accordance with the lever rule. It worth noting that a similar dependence was earlier reported for CoCrFeMnNiV_x HEAs composed of the fcc and σ phases [36,54]. Calculation of the bcc and σ phases fractions using the Al and Ni concentrations both in the respective phases and in the corresponding alloy as per data in Tables 2 and 3, gave the fractions of the bcc phase of 85, 70 and 15% in the x = 5, 7, and 10 alloys, respectively. The obtained values are reasonably close to the measured fractions (87, 68, and 23%, respectively; Table 3). A good correlation between the predicted and measured values confirms that the fraction of the phases in the Ti_(50-1.5625x)Nb_(30-0.9375x)Cr₁₀V₁₀Ni_{1.5x}Al_x alloys can be estimated using the lever rule. This gives the opportunity to tailor the structure and properties of the alloys by adjusting their chemical composition.

Meanwhile structure of the program alloys was found to be dependent on the annealing treatment also. The initial as-cast alloys consisted of the bcc phase with dendritic segregations (Fig. 3a, c, e). The interdendritic areas are expected to be enriched with "less refractory" elements (i.e. Al, Ni or Ti) like, for example, in the x = 0 alloy (Table 2). However, in the x = 5 alloy the interdendritic areas were rich in Ti but depleted of Al and Ni (Table 2). Most probably, this is the result of the Ni and Al partitioning between the σ and Ti₂Ni phases. It is reasonable to suggest that immediately after solidification the interdendritic areas were enriched with Al and Ni, but these elements were consumed during precipitation of the secondary phases through solid-state reactions. The morphology of the particles and the difference between their crystal structure and the structure of the "parent" bcc phase suggest that their nucleation was heterogeneous, and the growth occurred via a diffusioncontrolled mechanism predominantly inside interdendritic areas. The preferred location of the Ti2Ni particles suggests that they have a lower solvus temperature than that of the σ -phase and thus the Ti₂Ni phase formed later. Initial enrichment of the interdendritic areas with Al and Ni is probably the reason why they serve as the preferred places for the σ and Ti₂Ni phases formation. It appears that a similar scenario is proper for x = 7 alloy, but the interdendritic areas in this case were completely consumed by the σ and Ti₂Ni phases. Although the structure of the ascast x = 10 alloy is quite complex, the bcc phase had a dendritic morphology (Fig. 3h) which might be an indicator of a similar phase transformation sequence.

The annealing treatment had resulted in only moderate changes in microstructure (Fig. 5). An obvious and well-anticipated annealing effect in the $Ti_{50}Nb_{30}Cr_{10}V_{10}$ and x = 5 alloys was associated with the dendritic segregation disappearance (Fig. 5a and c). In addition, in the alloys containing the σ and Ti_2Ni phases (x = 5 and x = 7; Fig. 5c and e, respectively) the second phases particles became somewhat coarser and attained more regular shape due to coalescence process in comparison with the as-cast condition (Fig. 3c and e). It also appears that an increase in the fraction of the second phases (in the x = 7 alloy, for example) was mostly associated with the already existing particles growth rather than

with the precipitation of new finer particles. Meanwhile structure of the x=10 alloy (Fig. 5g) was even more stable than that of the rest of the alloys, most probably because of slow diffusion through the σ matrix.

Use of transition metals was recently proposed as a promising alloying strategy to improve properties of RHEA [55], yet it was never explored before. The current study suggests rather an unexpected result in the case of simultaneous usage of Ti (one of the most frequent elements of RHEAs [2]) and Ni so far. The coarse incoherent σ and Ti₂Ni particles appeared instead of anticipated, presumably fine and coherent B2 particles. Nevertheless, the obtained data shows that chemically complex RHEAs can contain phases that cannot be expected from the binary and ternary phase diagrams in contrast to some previous results [56]. This finding can open a new way to develop alloys with unexpected, novel phases.

4.2. Relationships between structure and mechanical properties

Mechanical properties at both room and elevated (800 °C) temperatures strongly depended on the chemical composition and microstructure of the $Ti_{(50-1.5625x)}Nb_{(30-0.9375x)}Cr_{10}V_{10}Ni_{1.5x}Al_x$ (x = 0, 5, 7, 10) alloys, (i.e. on the Ti, Nb, Ni, and Al concentration; Fig. 6, S1 and Table 4, S3). The most pronounced changes in the microstructure were related to the secondary σ and Ti_2Ni phases formation, and the amount of these phases and their morphology changed significantly with variations in the chemical composition. Since the σ and Ti₂ phases were much harder than the bcc matrix of the initial Ti₅₀Nb₃₀Cr₁₀V₁₀ alloy (Table 5), compression mechanical properties of the alloy were likely to be related to the fraction of secondary phases. Dependences of the yield strength at room temperature or at 800°C and compression ductility at room temperature on the fraction of the σ and Ti₂Ni phases are shown in Fig. 8. Note that (i) the second phases were mostly represented by the σ -phase with the ratio between the fractions of the Ti₂Ni and σ phases of \sim 1:5–1:10; (ii) ductility at 800°C are not plotted since all the alloys were compressed to the same strain level (>50%).

One can clearly see that indeed both strength and ductility of the alloys show pronounced dependence on the fraction of the σ and Ti_2Ni phases, yet the shape of these dependencies was different. For instance, at room temperature the yield strength demonstrated a sigmoidal type curve on the fraction of the phases (Fig. 8a). This type of dependence can be interpreted in the following way. When the fraction of the second phase particles is low (i.e. like 13-14% in the x = 5 alloy), they are rather coarse, many of them are located at the grain boundaries, and therefore they do not act as effective obstacles for the dislocations moving in the bcc matrix. Meanwhile when their fraction increases (to 28–32% in the x = 7 alloy), the coarse particles start to form a continuous network (Figs. 3f and 5f). In this case, they decrease the free dislocation path in the bcc matrix and therefore increased the strength of the alloys. Finally, when the fraction of the particles increases further (to 69-77% in the x = 10 alloy), weak dependence of strength on the fraction of the particles was observed. Note that the transition occurred

at the volume fraction of the second phases ~50%, i.e. when σ became the matrix phase. It is likely that deformation in this case is controlled by processes in the σ -phase, and the fraction of the softer bcc solid solution does not affect strength of the alloy. Note that a similar type of behavior was recently reported for Fe-Mn-Cr-Ni-Al TMHEAs composed of a softer fcc and harder bcc + B2 phases [33].

However, test points scattering in the case of a similar volume fraction of the second phases (evident at $V_f > 30\%$) in Fig. 8a implies that other factors also contribute to the properties. One of the factors is the particle morphology. For example, in the as-cast x = 7 alloy the bcc areas are often surrounded completely by the σ particles (Fig. 3f), thus limiting the possibilities to transfer slip from one bcc grain/area to another. Instead, in the same alloy in the annealed condition the σ -phase existed in the form of individual particles or complex shape conglomerates surrounded by the continuous bcc matrix. Most probably these differences were responsible for the higher strength of the as-cast alloy despite the similar fractions of the σ -phase in both conditions (Fig. 8a). Besides the morphology of the phases, the grains/particles size and the variations in chemistry (Figs. 3 and 5, Tables 3 and 5) can be responsible for differences in mechanical properties (strength and ductility at room and elevated temperatures (Fig. 8)) between the as-cast and annealed conditions. The detailed analysis is required to elucidate the exact factors which govern the mechanical behavior of each alloy, which is beyond the scope of the current paper, however.

The σ phase is known to be extremely brittle judging by the experience from conventional metallic alloys [57,58]. The available data on TMHEAs [59–61] and RHEAs [25] also confirmed a drastic decrease in ductility due to this phase. The results obtained in the current study agree with the previous reports; a pronounced decrease in ductility with an increase in the fraction of the second phases (mostly, σ) particles was observed (Fig. 8c). The dependence of ductility on the second phase particle fraction was not linear, however. It appears that at a relatively low σ -phase fraction when the ductile bcc solid solution phase was the matrix (x = 5–7 alloys), the σ particles served as a preferred point for the crack nucleation and propagation and therefore an increase in their fraction reduced ductility considerably. When the σ -phase became the matrix phase (x = 10 alloy), the alloy fractured already during the elastic part of the curve due to the high brittleness of the phase (Fig. 6a).

Finally, the strength of the alloys at 800°C demonstrated an almost linear dependence on the fraction of the second phases (Fig. 8b). This is most likely because of the significant softening of the bcc phase, which results in very low strength of both the $Ti_{50}Nb_{30}Cr_{10}V_{10}$ and x = 5 alloys (Fig. 6b). Only the presence of harder second phase particles with a significant fraction resulted in pronounced strengthening. However, even the x = 10 alloy with 60–65% of the σ -phase demonstrates only moderate strength of 280-285 MPa at 800 °C, which is quite low in regards to typical RHEAs properties [2]. The linear approximation of the available data suggests that the alloy fully composing of the σ (mostly) and Ti₂Ni phases should have strength of only \sim 350 MPa at 800 °C. This estimation shows that the σ -phase also loses its strength at 800 °C. The confirmation of this assumption can be found in the post-deformation microstructure of the x = 10 alloy (Fig. 7c), which shows the σ particles that clearly undergone plastic deformation. Gradual involvement of the σ -phase particles in plastic flow can result in a decrease in flow stress, observed at the stress-strain curves for the x = 10 alloy (Fig. 6b).

In summary, the obtained results suggest that the Ti, Ni-rich σ -phase strengthens the Ti_(50-1.5625x)Nb_(30-0.9375x)Cr₁₀V₁₀Ni_{1.5x}Al_x (x = 0, 5, 7, 10) alloys significantly at room temperature, yet the σ particles also results in a loss in ductility. Note that proper microstructure control via thermomechanical processing can most probably result in a more balanced combination of the properties [62–64]. Meanwhile, the σ phase particles do not provide sufficient strengthening at 800 °C. Further studies are required to establish the temperature interval where the Ti, Ni-rich σ -phase reinforced alloys retain their strength.

5. Conclusions

In this study, the as-cast and annealed structures of the $Ti_{(50-1.5625x)}Nb_{(30-0.9375x)}Cr_{10}V_{10}Ni_{1.5x}Al_x$ (x = 0, 5, 7, 10) refractory high entropy alloys and compression mechanical properties were analyzed and the following conclusions were drawn:

- 1) The $Ti_{50}Nb_{30}Cr_{10}V_{10}$ had a single bcc phase structure in the as-cast and annealed conditions and after deformation at 800°C in agreement with ThermoCalc predictions. The gradual replacement of Ti and Nb with Al and Ni resulted in the formation of the coarse Ti, Nirich σ and Ti₂Ni phases. The fraction of the σ -phase increased considerably from 11-13% in the x = 5 alloy to 60–65% in the x = 10 alloy while the respective fractions of Ti₂Ni were 1–2% and 9–12%, respectively. Annealing changed the morphology of the microstructure but barely affected the amounts of the constitutive phases.
- 2) The Ti, Ni-rich σ -phase formation was never reported before and it is not anticipated based both on the equilibrium phase diagrams obtained by the ThermoCalc software (which predicted the formation of a Ni-rich B2 phase) and corresponding binary and ternary phase diagrams. The composition of the σ particles can be described as (Ti, Cr, Nb, V)_{0.7}(Ni, Al)_{0.3}. The total concentration of Al and Ni in the bcc and σ phases remained almost constant in all the alloys; the fraction of the σ particles can thus be predicted using the lever rule.
- 3) The σ and Ti₂Ni phases formation considerably influenced the mechanical properties of the alloys. For example, at room temperature the yield strength increased from 755 MPa in the alloy with the single bcc phase structure (x = 0) to 1270 MPa in the alloy with 32% of the second phases (x = 7). At the same time, ductility decreased from >50% to 8.9%, respectively. The alloy with the σ -phase matrix (x = 10) fractured in the elastic region at room temperature. However, even the x = 10 alloy demonstrated the yield strength of only 285 MPa at 800 °C, while the other alloys were even softer.

Data availability

The raw/processed data required to reproduce these findings cannot be shared at this time as the data also forms part of an ongoing study.

Declaration of competing interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

CRediT authorship contribution statement

E.S. Panina: Investigation, Data curation, Visualization. N.Yu Yurchenko: Investigation, Visualization, Writing - review & editing. S. V. Zherebtsov: Writing - review & editing, Validation. M.A. Tikhonovsky: Writing - review & editing, Resources. M.V. Mishunin: Investigation. N.D. Stepanov: Conceptualization, Methodology, Supervision, Writing - original draft.

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Appendix A. Supplementary data

Supplementary data to this article can be found online at https://doi.org/10.1016/j.msea.2020.139409.

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